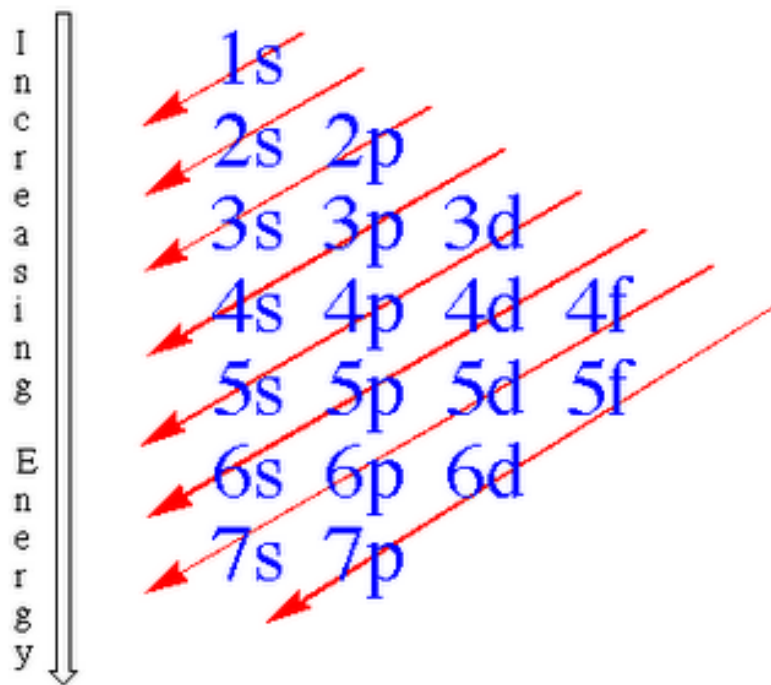


Sub-level	No. of electrons
<i>s</i> -	2
<i>p</i> -	6
<i>d</i> -	10
<i>f</i> -	14



1	s			
2	s	p		
3	s	p	d	
4	s	p	d	f
5	s	p	d	f
6	s	p	d	
7	s	p		
8	s			

Example: sulfur (S) with atomic number= 16

